



### 3. Results and discussion

The transmitted spectra of untreated DI water  $I_0$  and PAW  $I$  were measured between 190 and 340 nm. The transmitted spectra were converted into absorbance ( $Abs$ ) using an equation (1) below,

$$Abs = -\log\left(\frac{I}{I_0}\right). \quad (1)$$

The transmittance of PAW decreased across a specific UV range between 190 nm and 250 nm; no other intensity changes in the visible and near-infrared range between 400 and 900 nm were measured.

Fig. 2a shows a schematic of plasma jet irradiation for DI water (4 mL). Transmittance is decreased as a function of time as shown in Fig. 2b. Using eq. (1)

transmittance was converted into absorbance, we observed absorption spectra having a broad width of ~30 nm and the peak at around 206 nm. A close inspection of Fig. 2c shows there is a peak shift: the absorption peak wavelength became shorter from 207.4 nm to 205.8 nm as function of time and eventually, it became stable at the peak wavelength of 205.8 nm after 20 min exposure to the plasma jet irradiation. The absorbance is also increased as function of time. It should note that the negative absorbance of the case of the direct APPJ exposure (10 min) at short wavelength range below 200 nm was caused by helium gas flow reduced  $O_2$  (aq.) concentration in the liquid [8].

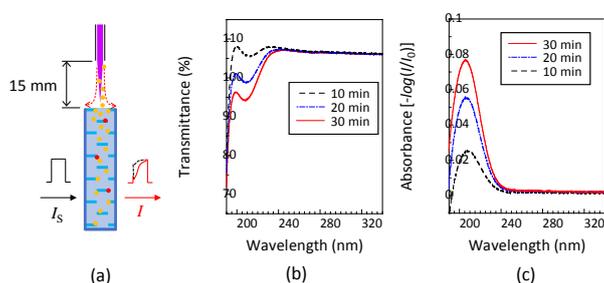


Fig. 2. (a) schematic of in-situ measurement of absorption spectra of DI water by a direct irradiation of APPJ, (b) transmittance of PAW as function of time and (c) converted absorbance from the transmittance.

Here we discuss the chemical composition of the PAW. As mentioned above, it is open reported that the long-living RONS typically  $H_2O_2$  and  $NO_x$  are in PAW. However there is no report of UV absorption spectroscopy at the short wavelength range. Thus we measured the absorption spectra for diluted  $H_2O_2$ ,  $HNO_3$ , and  $NaNO_2$ , respectively, with several different concentrations. Typical absorption spectra of the three reference solutions are shown in Fig. 3. We see the both  $NaNO_2$  and  $HNO_3$  have a broad width and absorption peaks between 200 and 210 nm, while  $H_2O_2$  has no peak in the region. The both absorption profiles of  $NaNO_2$  and  $HNO_3$  are very similar to the absorbance spectrum of PAW in Fig. 2c. However  $NO_x$  cannot clearly explain the absorption tail at longer wavelength over 250 nm,  $H_2O_2$  should be included in PAW. From the measurement of reference solutions, it is believed that the most RONS generated in the plasma-air interaction and transported by the gas flow [9].

Fig. 4a shows a simple experiment with a quartz barrier which placed on the top of DI water filled in the cuvette to block the gas flow but to transmit UV photons mainly related 2<sup>nd</sup> Positive  $N_2$ . Figs. 4b for the transmittance and 4c for absorbance show no significant absorption. The result indicated high energy UV photons play a minor role to generate reactive species in PAW. We also confirmed that there are no significant effect of UV photons using three standard  $D_2$ , Xe arc, and mercury lamps.

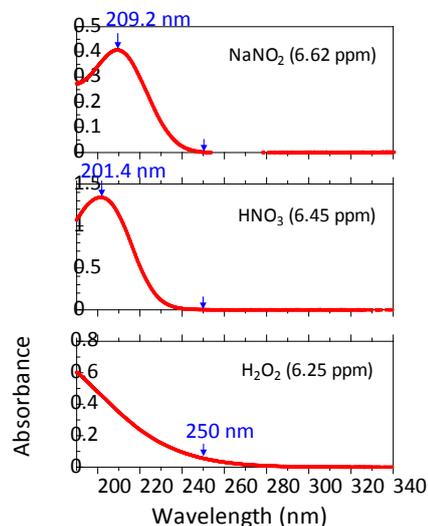


Fig. 3. Absorbance spectra of reference solutions (a)  $NaNO_2$  (6.62 ppm), (b)  $HNO_3$  (6.45 ppm), and (c)  $H_2O_2$  (6.25 ppm) diluted in DI water, respectively.

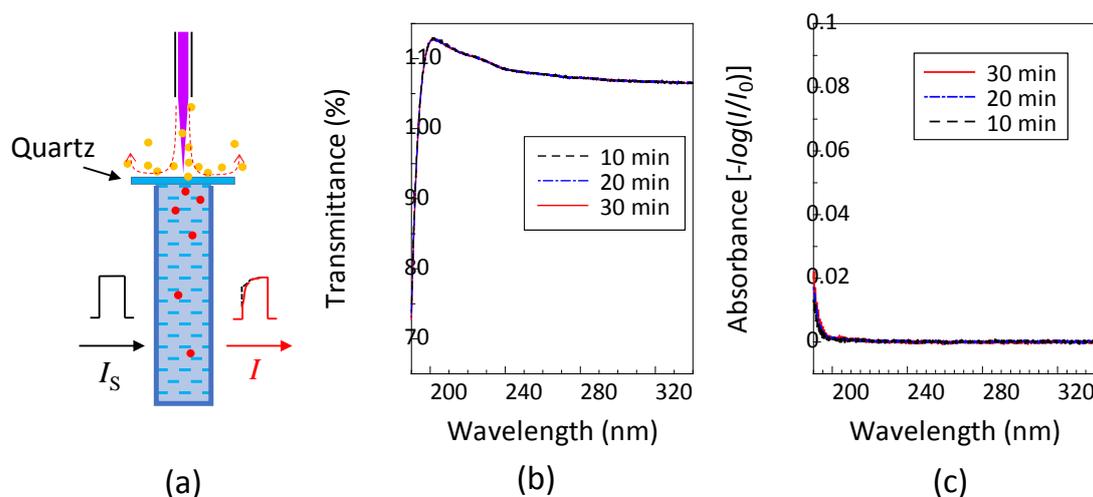


Fig. 4. (a) schematic of in-situ measurement of absorption spectra of DI water by an irradiation of APPJ onto a quartz plate, (b) transmittance of DI water as function of time and (c) converted absorbance from the transmittance.

#### 4. Conclusions

Using an in-situ absorption spectroscopy we successfully measured UV absorption of PAW induced by an APPJ irradiation. The absorption spectrum has a simple spectral profile indicated its simple chemical composition of  $\text{NO}_x$  and  $\text{H}_2\text{O}_2$ . Using a quartz barrier it remarked the RONS were generated by emerging plasma-air interaction and the He gas flow played a very important role in the transportation of plasma-generated RONS into DI water.

#### 5. Acknowledgements

This work was supported by the Priority Research Grant of KUT and partly by MEXT KAKENHI Grant-in-Aid for Challenging Exploratory Research (Grant number 26600129). We wish to thank Dr. Petr Lukeš at the Institute of Plasma Physics, Academy of Sciences of the Czech Republic and also Prof. Katsuhisa Kitano at Graduate School of Engineering, Osaka University for the valuable comments on the UV absorption spectra.

#### 6. References

- [1] P. Bruggeman and C. Leys. *J. Phys. D: Appl. Phys.*, **42**, 053001 (2009)
- [2] S. Ikawa, *et al.* *Plasma Process. Polymers*, **7**, 33-42 (2010)
- [3] P. Lukeš, *et al.* *Plasma Sources Sci. Technol.*, **23**, 015019 (2014)
- [4] A. Tani, *et al.* *Appl. Phys. Lett.*, **100**, 254103 (2012)
- [5] E. Takai, *et al.* *Plasma Process. Polymers*, **9**, 77-82 (2012)
- [6] K. Oehmigen, *et al.* *Plasma Process. Polymers*, **8**, 904-913 (2011)
- [7] K. Sakuramoto, *et al.* in: *Abstracts of 5th ICPM*. 166 (2014)
- [8] J.-S. Oh, *et al.* *Biointerphases*, submitted (2015)

- [9] J.-S. Oh, *et al.* *J. Phys. D: Appl. Phys.*, **44**, 155206 (2011)