

# IN SITU FOURIER TRANSFORM INFRARED ELLIPSOMETRY FOR MONITORING c-Si ETCHING PROCESS BY CF<sub>4</sub> PLASMA

T. Shirafuji, K. Tachibana<sup>†</sup> and S. Muraishi<sup>††</sup>

Department of Electronics and Information Science,  
Kyoto Institute of Technology, Matsugasaki, Sakyo-Ku, Kyoto 606, Japan

<sup>†</sup>Department of Electrical and Electronic Engineering, Kyoto University,  
Yoshida Hommachi, Sakyo-Ku, Kyoto 606, Japan

<sup>††</sup>Application and Research Center, JEOL  
1-2 Musashino 3 Chome, Akishima, Tokyo 196, Japan

## Abstract

A Fourier-transform infrared phase-modulated spectroscopic ellipsometry has been applied to *in situ* monitoring of c-Si surface treated by a parallel-plate RF plasma of CF<sub>4</sub> gas. Formation of fluorocarbon layer on the grounded electrode and fluorosilicon layer on the powered electrode have been identified from measured spectra.

## 1. Introduction

Eching of c-Si wafers in the plasma of F-containing gases has been studied with great deal of experimental and theoretical efforts, and basic mechanisms have been clarified on the basis of the results in the simplified conditions using XeF<sub>2</sub>[1] and F<sub>2</sub>[2]. Realistic RIE, however, mostly utilizes plasma of fluorocarbon gases[3]. Gas-phase and surface physics and chemistry become more complicated in this system due to the formation of fluorocarbon layer and surface roughness. Current activities for understanding the RIE are mostly aimed at gas-phase chemistry[4], and few attempts are made at surface where the etching proceeds actually. In this paper, we report on the results of *in situ* monitoring of c-Si surface in CF<sub>4</sub> plasma by using a FT-IR phase modulated spectroscopic ellipsometer (PMSE)[5].

## 2. Experimental

Figure 1 shows the schematic diagram of our FT-IR PMSE combined with a capacitively coupled RF plasma reactor for *in situ* surface investigation. The configuration consists of a conventional FT-IR spectrometer as a light source (JEOL JIR-7000), a polarizer, a photoelastic modulator (PEM) (Hinds PEM-90), a sample (c-Si wafer), folding mirrors M<sub>1</sub>-M<sub>2</sub>, an analyzer and a mercury-cadmium-telluride (MCT)

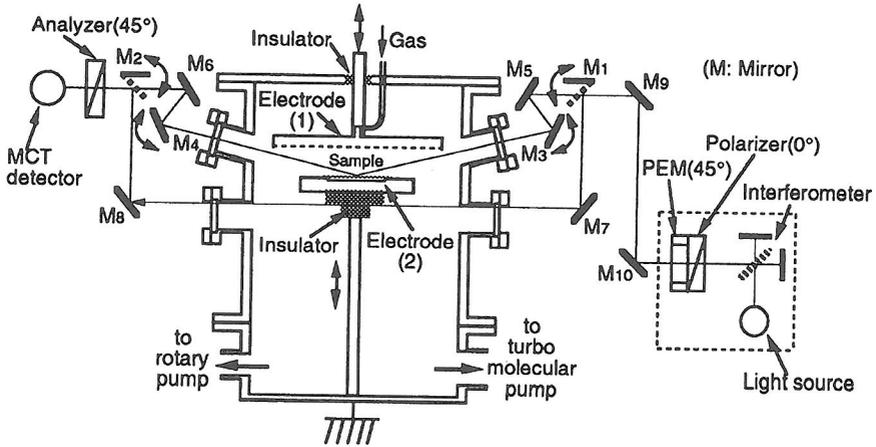


Fig.1 Experimental setup for an FT-IR PME system equipped to a conventional capacitively coupled RF plasma reactor.

detector cooled with liquid nitrogen. Modulation frequency ( $f$ ) of the PEM is 37kHz. The azimuth angle of the polarizer, modulator and analyzer are  $0^\circ$ ,  $45^\circ$  and  $45^\circ$ , respectively. Windows for the IR beam are made of KBr. The incidence angle of the IR beam on the wafer is  $75^\circ$ .

Spectral resolution was  $4\text{cm}^{-1}$ . Spectra of ellipsometric parameters  $\Psi$  and  $\Delta$  were calculated after 500-scans accumulation of  $dc$ ,  $f$  and  $2f$  signals detected through low-pass filter and lock-in amplifier. The accumulation took approximately 30 minutes. Optical density  $D$  was calculated with the following equation[5],

$$D = \text{Re}\{D\} + i \text{Im}\{D\} = \ln\left[\frac{\tan \Psi_{\text{sub}}}{\tan \Psi}\right] + i (\Delta_{\text{sub}} - \Delta)$$

In this equation, the subscript 'sub' means the value for the substrate before plasma treatment.  $\text{Re}\{D\}$  and  $\text{Im}\{D\}$  are proportional to imaginary and real part of dielectric function of surface over layer, and also to its thickness. Therefore,  $\text{Re}\{D\}$  and  $\text{Im}\{D\}$  have a Kramers-Kronig relationship, that is, a  $\text{Re}\{D\}$  peak appears as a gaussian peak and a  $\text{Im}\{D\}$  peak does as its differential.

In our setup, transmission absorption spectroscopy can also be performed by changing the angle of the two mirrors,  $M_1$  and  $M_2$ . The IR beam for absorption spectroscopy passes just above the wafer surface, and the distance between the beam and the surface is approximately 5mm. Pass length in the chamber is approximately 30cm. Absorbance spectrum has been calculated by subtracting the spectrum taken under evacuated condition. For the purpose of relative comparison, the spectrum has been normalized with the absorbance value of  $\text{CF}_4$  peak at  $1283\text{cm}^{-1}$  for pure  $\text{CF}_4$  gas without discharge. The peaks have been assigned according to ref.6.

$\text{CF}_4$  plasma treatment was carried out on c-Si(100) wafer ( $5 \times 5\text{cm}^2$ ). Temperature of the wafer was not controlled intentionally. The back surface of the wafer was

roughened, and it was separated from the holder surface by 5mm in order to reduce the interference in the spectrum caused by multiple reflection. The diameter of RF electrode was 16cm. Flow rate, pressure and RF power were 8sccm, 0.2Torr and 100W, respectively. The PMSE measurement was carried out under evacuated condition after each CF<sub>4</sub> plasma treatment.

### 3. Results and Discussion

#### 3.1 Gas-phase diagnostics

Figure 2 shows IR transmission absorbance spectra for CF<sub>4</sub> plasma taken above a c-Si wafer on the grounded (GND) and powered (RF) electrodes and for CF<sub>4</sub> gas without discharge. In the present experimental condition, dissociation degree of CF<sub>4</sub> is found to be 20-30% by comparing the height of CF<sub>4</sub> peak at 1283cm<sup>-1</sup>. As the species generated through the dissociation of CF<sub>4</sub>, CF<sub>3</sub> and CF<sub>2</sub> are detected at 1250cm<sup>-1</sup> and 1115cm<sup>-1</sup>, respectively. CF peak at 1279cm<sup>-1</sup> cannot be distinguished from CF<sub>4</sub> peak. In addition to these fluorocarbon peaks, SiF<sub>4</sub> peak appears at 1028cm<sup>-1</sup> due to etching of c-Si wafer. The peak height taken above the RF electrode is slightly higher than that above the GND electrode. Although diffusional transport occurs in gas phase, this shows higher etching rate on the RF electrode. Although SiF<sub>2</sub> is known to be an alternative etching product, it is not identified in the corresponding band around 850cm<sup>-1</sup>.

Density of these species are summarized in Table I. The density of CF<sub>2</sub> and CF<sub>3</sub> has been calculated from integrated absorption coefficient for these peaks[7]. The density of SiF<sub>4</sub> has been calculated from peak absorption coefficient with a

Table I. Estimated density of CF<sub>4</sub>, CF<sub>3</sub>, CF<sub>2</sub> and SiF<sub>4</sub> (in cm<sup>-3</sup>).

RF	CF <sub>4</sub>	CF <sub>3</sub>	CF <sub>2</sub>	SiF <sub>4</sub>	Sum
0W	6.44x10 <sup>15</sup>	0	0	0	6.44x10 <sup>15</sup>
100W	4.71x10 <sup>15</sup>	5.40x10 <sup>14</sup>	7.50x10 <sup>14</sup>	2.79x10 <sup>14</sup>	6.28x10 <sup>15</sup>

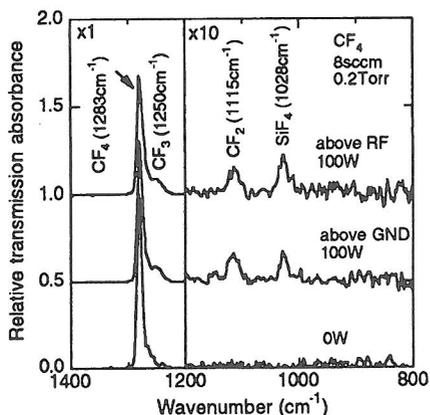


Fig.2 IR transmission spectra of CF<sub>4</sub> plasmas.

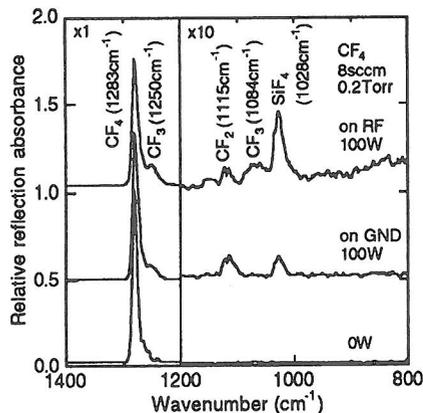


Fig.3 IR reflectance spectra of CF<sub>4</sub> plasmas.

proportionality constant of  $3.53 \times 10^{-18} \text{cm}^2$ , which has been determined by measuring absorption coefficient for pure  $\text{SiF}_4$  gas under known pressure. As can be seen in the table, density of  $\text{CF}_2$ ,  $\text{CF}_3$  and  $\text{SiF}_4$  is about 10% of  $\text{CF}_4$ . In order to estimate the upper limit of  $\text{SiF}_2$  density, we have tried a measurement in  $\text{SiF}_4$  plasma. However,  $\text{SiF}_2$  peak cannot be detected under the condition of 0.2 Torr and 100 W. A simulation of gas phase reactions in  $\text{SiF}_4$  plasma using a one dimensional fluid model predicts  $\text{SiF}_2$  density is in the order of  $10^{10} \text{cm}^{-3}$ . Therefore, the density of  $\text{SiF}_2$  in  $\text{CF}_4$  plasma is regarded to be less than  $10^9$ - $10^{10} \text{cm}^{-3}$ . From these results, we can conclude that major etching product in the gas phase is  $\text{SiF}_4$  in  $\text{CF}_4$  plasma.

Figure 3 shows IR reflection absorbance spectra of c-Si wafer in  $\text{CF}_4$  plasma. As we have not subtract the gas-phase absorption, the spectra contain both gas phase and surface contribution. If the surface (or near surface) absorption is negligible, the spectra in Fig.3 must be similar to those in Fig.2 obtained in the gas phase. Comparing these figures, we can see that the spectrum reflected on the RF electrode shows remarkably differs from that on the GND electrode. In the former,  $\text{CF}_3$  peak at  $1250 \text{cm}^{-1}$  shows higher intensity than in the gas phase spectrum, and another clear peak assigned to  $\text{CF}_3$  appears at  $1084 \text{cm}^{-1}$ , which cannot be identified in Fig.2. In addition,  $\text{SiF}_4$  peak shows much higher intensity. These differences show that RAS spectra contain the surface absorption,

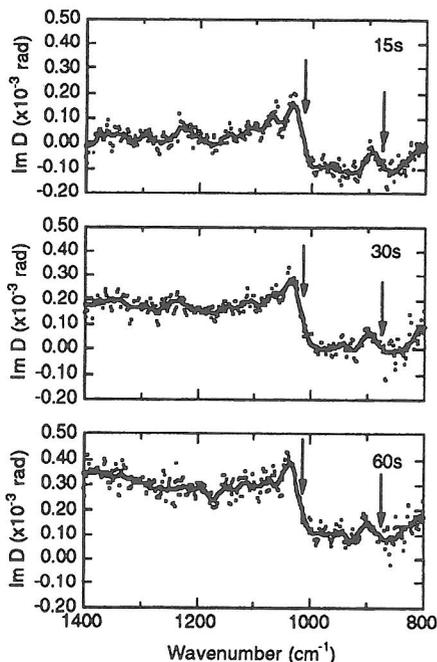


Fig.4 Temporal behavior of  $\text{Im}\{D\}$  spectra for c-Si surface treated with  $\text{CF}_4$  plasma on RF electrode.

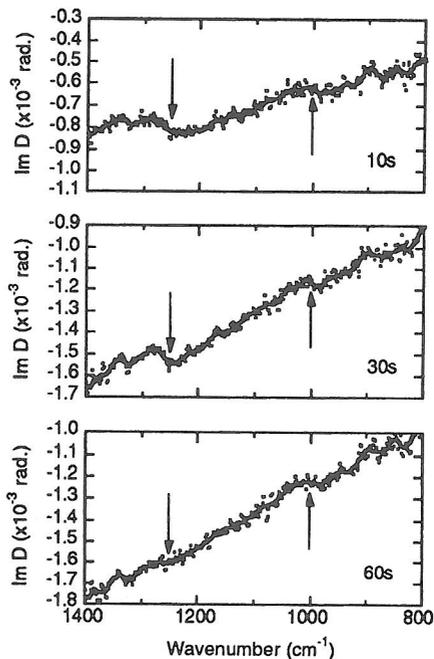


Fig.5 Temporal behavior of  $\text{Im}\{D\}$  spectra for c-Si surface treated with  $\text{CF}_4$  plasma on GND electrode.

and the difference between the RF side and the GND side is attributed to the difference in the self bias. Higher intensity of SiF<sub>4</sub> peak on the RF side is due to higher rate of c-Si etching, and the production of SiF<sub>4</sub> become larger. Higher intensity of CF<sub>3</sub> peak may be due to its production in the etching of polymerized CF<sub>2</sub> on the surface.

### 3.2 Surface diagnostics

Figure 4 shows spectra of Im{*D*} for c-Si surface treated with CF<sub>4</sub> plasma on the RF electrode for 15, 30 and 60 seconds. Etching rate was 3-5Å/s, which was determined in the etching of poly-silicon film formed on quartz substrate by spectroscopic ellipsometry in visible region. In the figure, two differential peaks appear at approximately 1010 and 870cm<sup>-1</sup>, which are assigned to SiF<sub>4</sub> and SiF<sub>2</sub>, respectively. As the major gas phase species is SiF<sub>4</sub>, SiF<sub>2</sub> formed on the surface does not desorb as it is, and it must turn into SiF<sub>4</sub> on the surface. Appearance of the SiF<sub>4</sub> peak suggests that SiF<sub>4</sub> molecules produced through the etching process still remain on the substrate surface although this species is volatile. The peak intensity and the absolute value of Im{*D*} does not change so much with increasing the treatment time. This means that the thickness of the overlayer (presumably fluorinated layer) does not change during etching. This tendency is similar to the result obtained in spontaneous etching with XeF<sub>2</sub>[1], although the existence of SiF<sub>4</sub> is not identified.

Figure 5 shows Im{*D*} spectra for c-Si surface treated on the GND electrode. SiF<sub>4</sub> peak appears at 1010cm<sup>-1</sup>, and SiF<sub>2</sub> peak also appears around 870cm<sup>-1</sup> although it is not so clear as in the case of RF electrode. In addition, a peak at 1250cm<sup>-1</sup> appears, and it is assigned to CF<sub>3</sub>. In this case, the base line of Im{*D*} spectra changes more rapidly than that in Fig.4. This means that relatively thick fluorocarbon film is formed on the surface.

In order to verify the peak assignment, we have measured the c-Si surface exposed to F<sub>2</sub>. Figure 6 shows Im{*D*} spectra of the surface exposed to 5%F<sub>2</sub> diluted with He at 0.5Torr for 5min. As seen in the figure, a peak is found at 850cm<sup>-1</sup>, which corresponding to SiF<sub>2</sub>. This means F<sub>2</sub> molecules adsorb dissociatively on the surface to forms SiF<sub>2</sub> bonds. Figure 7

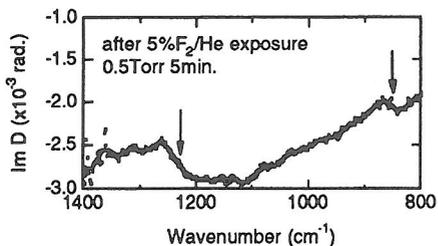


Fig.6 Im{*D*} spectrum for c-Si surface exposed to 5%F<sub>2</sub>/He for 5 minutes.

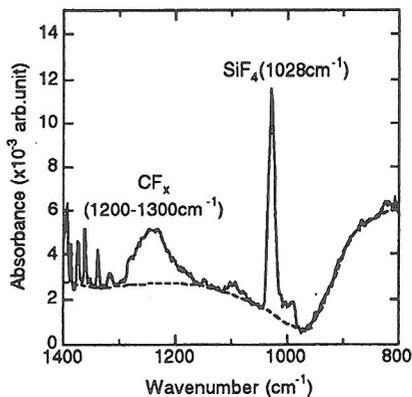


Fig.7 Transmission absorbance of gas phase during F<sub>2</sub>-exposure experiment.

shows transmission absorption spectrum in gas phase during F<sub>2</sub> exposure, in which a SiF<sub>4</sub> peak clearly appears at 1028cm<sup>-1</sup>. Therefore, F<sub>2</sub> molecules form SiF<sub>2</sub> bonds on the surface, and the SiF<sub>2</sub> bonds desorb out as SiF<sub>4</sub> molecules. However, SiF<sub>4</sub> is not found in the surface spectrum, while it is found for CF<sub>4</sub> plasma treatment. Therefore, remaining SiF<sub>4</sub> on the surface treated with CF<sub>4</sub> plasma is regarded to be piled in fluorinated layer. Although a broad peak around 1250cm<sup>-1</sup> corresponding to fluorocarbon is found in Fig.7 even though no fluorocarbon gas is used, this is considered to be due to removal of fluorocarbon film from chamber wall and re-deposition on the surface.

## 5. Conclusion

In conclusion, an FT-IR PMSE has been applied to monitoring of c-Si surface treated with CF<sub>4</sub> plasma. Gas-phase transmission absorption spectroscopy and simple reflection spectroscopy have also been carried out.

Gas-phase absorption spectroscopy has revealed that major etching product in gas phase is SiF<sub>4</sub>, and the density of SiF<sub>2</sub> is much lower.

PMSE measurement has shown that SiF<sub>4</sub> and SiF<sub>2</sub> exist on the surface after the treatment on the RF electrode, while CF<sub>x</sub> also exist on the GND electrode. An F<sub>2</sub> exposure experiment has shown that SiF<sub>4</sub> is dominant in gas phase and only SiF<sub>2</sub> exists on the surface, which suggests that SiF<sub>2</sub> turns into SiF<sub>4</sub> to desorb out. Pile up of SiF<sub>4</sub> in the case of CF<sub>4</sub> treatment is considered to be peculiar to this treatment.

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