

PLASMA ETCHING OF Si BY SF<sub>6</sub>, CF<sub>4</sub>, AND SF<sub>6</sub>/H<sub>2</sub> AND CF<sub>4</sub>/H<sub>2</sub> MIXTURES

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Keywords: Plasma, EtchingCompounds: Silicon, Sulfur hexafluoride, Carbon tetrafluorideABSTRACT

Silicon chips were etched in D.C. glow discharges in SF<sub>6</sub>, CF<sub>4</sub> and in their mixtures. Steady state mass spectrometric signals were measured as a function of the power input. At higher powers, subsequent exposure of the Si to H<sub>2</sub> discharges led to the release of copious amounts of SiF<sub>4</sub> and H<sub>2</sub>O. Auger analysis showed S, O and F but no surface Si. CF<sub>4</sub> studied under comparable conditions was found to be far less reactive toward Si.

1. INTRODUCTION

Reactive ion etching of Si in a CF<sub>4</sub> plasma has been studied by many authors (1,2), and recognizing the complexity of this process one should certainly seek to determine the way the overall etch rate and the composition of the plasma depend on several independent parameters, if one wishes to improve our understanding of the individual processes involved.

The present study concentrates on a comparison of the SF<sub>6</sub>/Si and CF<sub>4</sub>/Si systems, both being studied under similar conditions. Since CF<sub>4</sub> is known to lead to rather tenacious C deposits on Si, it seems very interesting to determine whether any S containing surface layers are formed during SF<sub>6</sub> etching which would perhaps be much more volatile and thus far less detrimental to the etching process. If this is so, one might further expect SF<sub>6</sub> to show little selectivity when comparing the etch rates of SiO<sub>2</sub> and Si. In addition to comparing these chemical systems, the input power constituted another important independent variable in this work, since early exploratory experiments had shown that the relative amounts of long lived species present in such plasmas strongly depend on the input power levels used.

2. EXPERIMENTAL

The discharge vessel consisted of Pyrex tubing, 25mm I.D. and 32cm long, at one end closed by a large tapered joint cap with various electric feedthroughs. The other end of this tube was closed by a flat window

with a 0.50mm aperture in its center. This opening led to a differentially pumped region and thence, via a second aperture of 1.59mm I.D., straight to the point of intersection with the electron beam of the ion source of a mass spectrometer.

The medium resolution instrument used was a Hitachi Model RMU-6E with variable entrance and exit slits and electron multiplier detector.

Two parallel stainless steel plate electrodes of  $2\text{cm}^2$  area were mounted approximately 5mm from the first aperture and 1.0cm apart.

Typical pressures in the plasma cell were 0.110 Torr, in the differentially pumped region  $1.6 \times 10^{-5}$  Torr, and in the mass spectrometer  $2 \times 10^{-6}$  Torr. The path length from the first aperture to the electron beam was about 7.0cm.

Using several needle valves, the etchant gas was admitted to the plasma vessel at a constant low rate of 0.02sccm, as measured by a Hastings Model ALL-5 flow meter, using the published corrections applicable to each etchant gas, or gas mixture. The discharge cell and electrode surfaces were cleaned before each series of experiments, using dilute HCl and Ar discharges.

Samples to be etched were centered on the lower electrode which was in most cases the cathode. The polycrystalline silicon samples were 99.999% pure, as obtained from the Apache Chemical Company. Etchant gases came from the Linde Division of Union Carbide and were of the following purities:  $\text{SF}_6$  - 99.9 mole %;  $\text{CF}_4$  - 99.7 mole %;  $\text{H}_2$  - 99.95 mole %.

After the  $\text{SF}_6$ ,  $\text{CF}_4$ , or  $\text{H}_2$  mixture plasma treatments, Ar and  $\text{H}_2$  discharges were used in order to remove various products which might have been deposited on the sample surface during the etching experiments.

The mass spectrometric signals were corrected for minor instrumental variations, using several internal standards which were known to be unaffected by the plasma chemical changes.

### 3. RESULTS

Figure 1 shows the results of a series of blank experiments using  $\text{SF}_6$ , and Figure 2 the corresponding results for the  $\text{SF}_6/\text{Si}$  etching experiments. A slight change of the mass spectrometer sensitivity which occurred between both series of experiments was obvious when comparing the  $\text{SF}_6$  cracking patterns at zero power input, that is the points on the ordinates of both figures. All data of Figure 1 were corrected for this shift.

Practically all the ionic species detected mass spectrometrically were created from neutrals effusing from the plasma cell, with only extremely small contributions from ions formed in the plasma.

To demonstrate more clearly the effect of the  $\text{SF}_6$  plasma upon the Si sample, the results of Figure 1 were subtracted from those of Figure 2, and the differences plotted in Figure 3.

The  $\text{SiF}_3^+$  signals from  $\text{SiF}_4$ , which resulted from the etching of the internal Pyrex surfaces were at least 20 times smaller than the  $\text{SiF}_3^+$  signals coming from the Si sample. For all power inputs, the steady state  $\text{SiF}_4$  cracking patterns were constant.

The  $\text{SOF}_2^+$  peaks, which contain oxygen originating from the internal walls also are unimportant compared to the  $\text{SiF}_3^+$  signals related to the sample.  $\text{SOF}^+$  and  $\text{SO}^+$  signals were also measured but were not presented in these figures because they are proportional to the  $\text{SOF}_2^+$  peaks and can therefore be considered part of the same cracking pattern.

By contrast, the  $\text{S}_2\text{F}_2^+$  peaks (and the related  $\text{S}_2\text{F}^+$  signals which are not shown) are probably formed from  $\text{SF}_6$  fragments, and constitute the only important by-product from that gas.

The electrode on which the sample was placed was negative with respect to the other electrode for all data presented above. For this polarity, etching was always distinctly faster than for the reverse polarity, especially at higher input power levels.

The  $\text{CF}_4/\text{Si}$  etching and  $\text{CF}_4$  blank results analogous to the  $\text{SF}_6/\text{Si}$  and  $\text{SF}_6$  blank data were obtained under identical conditions but are not presented in detail because they are consistent with results previously published by others. It was observed, however, that the  $\text{CF}_4$  etching rates were far lower than those of  $\text{SF}_6$ , leading to a much more pronounced wall effect in the present apparatus. Also large quantities of oligomers were formed, that is  $\text{C}_2\text{F}_6$ ,  $\text{C}_2\text{F}_4$ , and  $\text{C}_3\text{F}_6$ , in that order. Low etching rates of Si under low  $\text{CF}_4$  flow rates have been observed in the literature (3).

The plasma cell pressure was observed to drop right after the discharge was established, it went through a minimum and then increased beyond its initial value, both in  $\text{SF}_6$  and  $\text{CF}_4$  experiments. This led to a few additional, exploratory experiments which will be mentioned below, in the Discussion.

#### 4. DISCUSSION

The mass spectrometric cracking pattern of  $\text{SF}_6$  as obtained in this work can be read from the points on the ordinates of Figures 1 and 2. The principal peak of  $\text{SF}_6$ , that is  $\text{SF}_5^+$ , drastically decreased with increasing power density, in the blank experiment, showing that  $\text{SF}_6$  is readily dissociated (Figure 1). When the sample is present, the  $\text{SF}_6$  is consumed at an even faster rate, especially at low input power, resulting in a deep minimum in the difference plot (Figure 3).

It was recognized very early that at least part of the  $\text{SF}_x^+$  signals ( $x=1$  to 5) must be due to the fragmentation of  $\text{SF}_6$  in the mass spectrometer ion source, while another major contribution originated from the  $\text{SF}_x$  radicals ( $x=1$  to 5) travelling from the plasma into the ion source. To obtain a closer estimate of this latter contribution to the  $\text{SF}_x^+$  signal ( $x=1$  to 4), it was assumed that all the  $\text{SF}_5^+$  seen in the mass spectra could be ascribed to  $\text{SF}_6$  molecules reaching the ion source, and the  $\text{SF}_x^+$  signals ( $x=1$  to 4) were corrected for these contributions.

Next, the cracking pattern corrected peaks (4) were used again to calculate corrected difference curves, shown as dashed curves in Figure 3, except for the  $\text{SF}_4^+$  differences, which were extremely low. The  $\text{SF}_3^+$ ,  $\text{SF}_2^+$ , and  $\text{SF}^+$  difference signals, when corrected in this fashion, are seen to peak at successively higher power levels but it must be added

that the reaction steps responsible for the increase of several species at even higher input powers can not as yet be characterized, from the present data.

$S_2F_2^+$ , the most important by-product in these experiments, peaks at a much higher input power than  $SF^+$ , indicating that the supply of  $SF^+$ , a possible precursor of  $S_2F_2$  is not the only important factor. Either the consumption of  $S_2F_2$  varies greatly with input power, or other precursors are to be considered.

As mentioned above, it seemed as if some plasma generated species might be depositing on the sample surface. To explore this, it was attempted to clean the etched samples by Ar,  $O_2$ , and  $H_2$  discharges. Ar and  $O_2$  plasma treatments produced only small amounts of released gases, but the  $H_2$  plasma produced large amounts of  $SiF_3^+$  and  $H_2O^+$ , plus some  $H_2S$ . Auger spectra showed large amounts of S, O, and F to be present in the surface, but no Si, before or after the  $H_2$  plasma treatment.

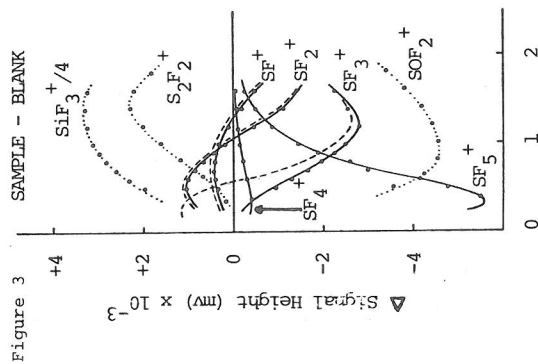
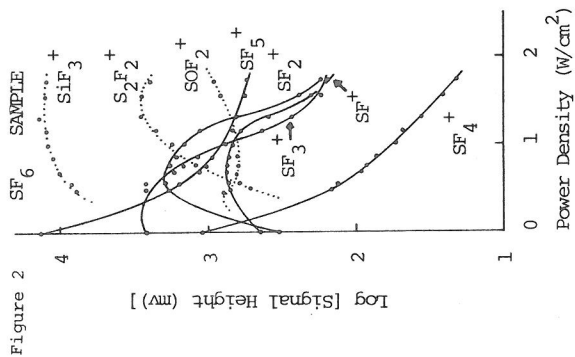
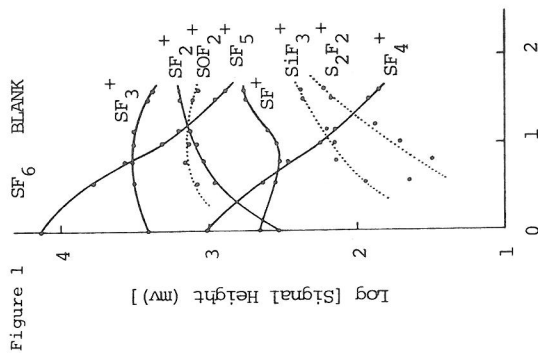
One might ask whether the efficient etching by  $SF_6$  can occur simultaneously with the improved release of  $SiF_4$  as induced by  $H_2$ , but the etching rate of  $SF_6/H_2$  mixtures was found to be lower than what corresponded to the  $SF_6$  mole fraction. One can thus conclude that the scavenging effect of  $H_2$  is far more detrimental than its facilitating of the  $SiF_4$  release. The production of HF was correspondingly high in these mixture experiments.

Returning briefly to the  $CF_4/Si$  results, it was found that the  $C_2$  and  $C_3$  fluorinated species formed, very much obliterated the  $CF_3^+$ ,  $CF_2^+$ , and  $CF^+$  peaks present, preventing the analysis of possible trends analogous to those found for the  $SF_x^+$  species ( $x=1$  to 3).

#### REFERENCES

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- (3) B. N. Chapman, V. J. Minkiewicz, J. Vac. Sci. Technol., 15, 329 (1978).
- (4) The cracking pattern corrected curves are not shown in Figures 1 and 2, for the sake of clarity, but copies of these drawings can be obtained by writing to the author.

Discharge Cell Pressure - 91 mtorr  $\text{SF}_6$   
 $\text{SF}_6$  Flow Rate - 0.02 sccm



Figures 1 and 2 - Log Mass Spectrometer Signal Height vs. Power Density for  $\text{SF}_6$  Blank (Fig. 1) and  $\text{SF}_6/\text{Si}$  (Fig. 2) Experiments.

Figure 3 -  $\Delta$  Mass Spectrometer Signal Height vs. Power Density for Sample (Fig. 2) - Blank (Fig. 1). Dashed curves show the cracking pattern corrections for the  $\text{SF}_x$  fragments (see text).